



Starch Update 2011 The 6th International Conference on Starch Technology

13 - 14 February 2012 Centara Grand at Central Plaza Ladprao Bangkok, Thailand

Organized by

National Center for Genetic Engineering and Biotechnology (BIOTEC)
National Science and Technology Development Agency (NSTDA)
Ministry of Science and Technology (MOST)
Kasetsart Agricultural and Agro-Industrial Product Improvement Institute (KAPI)
Kasetsart University (KU)

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Ministry of Science and Technology

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Starch Update 2011: The 6th International Conference on Starch Technology Monday 13 - Tuesday 14 February 2012 Centara Grand at Central Plaza Ladprao Bangkok, Thailand

Monday 13 February 2012

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THE RESERVE AS A STOCK OF THE PROPERTY AS A STOCK OF	- Welcome remarks
	By Assoc. Prof. Dr. Klanarong Sriroth
	Chairman of organizing committee
>- 1	- Opening remarks
	By Dr. Kanyawim Kirtikara
08:30 – 09:00	Executive Director of National Center for Genetic Engineering and Biotechnology (BIOTEC)
	Ms. Yupa Pankaew
	Director of Kasetsart Agricultural and Agro-Industrial Product Improvement Institute (KAPI)
	Plenary lecture 1: "Insights into starch granule swelling and solubility"
09:00 - 09:30	By Prof. Sandra Hill
	The University of Nottingham, UK
	Plenary lecture 2: "Designed structure and digestibility of A- and B-type crystals
00-20 10-00	through controlled assembly of short-chain amylose"
09:30 - 10:00	By Prof. Yong-Cheng Shi
	Kansas State University, USA
10:00 = 10:30	Coffee break
	Plenary lecture 3: "From blocklets to granule: exploring starch granular architecture"
10:30 - 11:00	By Prof. Dr. Koushik Seetharaman
	University of Guelph, Canada
	Plenary lecture 4: "How internal chain structure of amylopectin might drive function"
11:00 - 11:30	By Prof. Dr. Bruce Hamaker
	Whistler Center for Carbohydrate Research, Purdue University, USA
	Plenary lecture 5: "Starch business in China"
11:30 – 12:00	By Jin Shu Ren
A	Starch Association of China and Sugar-Alcohol Association of China, P.R. China
12:00 – 12:20	Industrial presentation: "Enzyme application for cassava starch extraction"
	By Ying Qian
1,00	Genencor, P.R.China
12:20 - 13:30	Lunch at Ladprao Suite
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	Gondurent Seribit (C	Smilns oral presentation)
	Session I (Ballroom A)	Session II (Ballroom B)
13:30 – 13:55	Oral presentation 1: "China starch industry development & modern processing technology" By Wang Yanbo Henan University of Technology, P.R. China	Oral presentation 11: "Acetylation of sago starch in densified CO ₂ " By Henky Muljana Parahyangan Catholic University, Indonesia
13:55 – 14:20	Oral presentation 2: "Effects of cassava crush size and variety on starch extraction efficiency" By Ruenrom Lerdlattaporn King Mongkut University of Technology Thonburi, Thailand	Oral presentation 12: "Silver particles prepared in chemical reduction of silver nitrate using hydroxypropyl starch" By Kritapas Laohhasurayothin National Science and Technology Development Agency, Thailand
14:20 – 14:45	Oral presentation 3: "Environmental impacts and benefits of cassava starch production using Life Cycle Assessment (LCA) thinking" By Dr. Thierry Tran CIRAD, France	Oral presentation 13: "Cassava starch nanocrystals with hydrophobically surface modification" By Kittiwut Kasemwong National Science and Technology Development Agency, Thailand
14:45 – 15:10	Oral presentation 4: "Atmospheric plasma: A non-chemical route for promoting cross-linking in starch" By Somsak Dangtip Mahidol University, Thailand	Oral presentation 14: "Effect of annealing treatments on the properties of flour from two different pre-germinated brown rice varieties" By Onamon Jungsrimsirisakhul Kasetsert University, Thailand
15:10 - 15:40	Coffee break	
15:40 – 16:05	Oral presentation 5: "The study of surface pores and channels of rice starch granules" By Piyada Achayuthakan Suan Sunandha Rajabhat University, Thailand	Orul presentation 15: "Effect of annealing on thermal and pasting properties of acidmethanol hydrolysed rice starch" By Junwadee Aryupong Kasetsart University, Thailand
16:05 – 16:30	Oral presentation 6: "Functional properties of Thai lotus seed (Nulumbo nucifera Gertn.) flours" By Urai Meesit Ubon Ratchathani University, Thailand	Oral presentation 16: "Improving the texture of frozen cooked rice (Oryza sativa L.) cv. Khao Dawk Mali 105" By Pornpisa Prasertthaijaroen Kusetsart University, Thailand

	Session I (Ballroom A)	Session II (Ballroom B)
16:30 – 16:55	Oral presentation 7: "Biobutanol production via Acetone-Butanol-Ethanol (ABE) fermentation of cassava chips by Clostridium Saccharoperbutylacetonicum DSM 14923" By Son Chu-Ky Hanoi University of Science and Technology, Vietnam	Oral presentation 17: "The effects of additives and white egg on composite cassava – wheat bread" By Judy Witono Parahyangan Catholic University, Indonesia
16:55 – 17:20	Oral presentation 8: "Potential of bioethanol production from inedible banana" By Sutera Srichalyos Mahasarakham University, Thailand	Oral presentation 18: "Caramel sauces thickened with combinations of starch and xanthan gum" By Magdalena Krystyjan University of Agriculture in Krakow, Poland
17:20 – 17:45	Oral presentation 9: "Characteristics of enzymatic hydrolysates from butyrylated arrowroot starch" By Hary Haryadi Gadjah Mada, Indonesia	Oral presentation 19: "Resistant starch composition and physicochemical properties of four banana cultivars during ripening" By Nednapis Vatanasuchart Institute of Food Research and Product Development, Kasetsart University, Thailand
17:45 – 18:10	Oral presentation 10: "Effects of pullulanase and isoamylase and rice varieties on resistant starch type III formation" By Ariya Lapkhoksung Suranaree University of Technology, Thailand	Oral presentation 20: "Oxidation of cassava starch using H ₂ O ₂ and UV-C irradiation to improve the baking properties" By Niken Palupi Jember University, Indonesia

Tuesday 14 February 2012

University of Guelph, Canada Spectacular lecture: "A Perspective on the Discoveries and Paradigm Shifts in Starch Research" By Prof. Dr. Koushik Seetharaman and Dr. Eric Bertoft By Prof. Dr. Koushik Seetharaman and Dr. Eric Bertoft By Prof. Dr. Koushik Seetharaman and Dr. Eric Bertoft By Prof. Dr. Koushik Seetharaman and Dr. Eric Bertoft By Prof. Dr. Koushik Seetharaman and Dr. Eric Bertoft By Prof. Dr. Koushik Seetharaman and Dr. Eric Bertoft By Prof. Dr. Koushik Seetharaman and Dr. Eric Bertoft By Prof. Dr. Koushik Seetharaman and Dr. Eric Bertoft By Eric Brown Seetharaman and Dr. Eric Bertoft By Takahiro Noda National University By Eric Brown Seetharaman and Dr. Eric Bertoft By Takahiro Noda By Macie Fledorowicz University of Agriculture Research Center for Hokkaido Region, Japan By Carl Fledorowicz University of Agriculture, Poland Oral presentation 22: "Comparison of molecular structure and selected physicochemical properties of spelt wheat and common wheat starches illuminate with visible polarized light and UV radiation" By Kim Hyun-Joong Seoul National University, South Korea 11:15-12:00	deside announce	Prising a Billionnell
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Oxidation of cassava starch using II2O2 and UV-C irradiation to improve the baking properties

*Niken W. Palupi(1), Eni Harmayani(2) and Haryadi(2)

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Abstract

Abstract
The aim of this study was to investigate the effect of oxidation using H₂O₂ and UV_C The aim of this study was to investigate the irradiation on the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content, pH as well as baking expansion and swelling solubility of the carboxyl content. the cassava starch. Commercial cassava starch was oxidized with 0.4% (v/v) H₂O₂ (30%) the cassava starch was oben solution) and irradiated with UV-C for 0.25 – 15 min. The treated starch was characterized starch was characterized and solubility. The for carboxyl content, baking expansion ability, swelling and solubility. The untreated commercial cassava starch was used as control. The result indicated that the cassava starch oxidized with H₂O₂ and UV-C irradiated for 3 min had the highest baking expansion (10.75) mL/g) and lower swelling and solubility compared to the native starch. The oxidized starch contained 0.36% carboxyl groups. The oxidized starch was then applied to substitute wheat flour to prepare bread. Bread made from 50% the oxidized cassava starch had a high volume good texture, and great appearance compared with wheat bread and native cassava starch bread.

Keywords: Oxidized cassava starch, baking expansion, UV-C irradiated, swelling and solubility

1. Introduction

Sun dried, fermented cassava starch or cassava sour starch, known as polvilho azedo in Brazil and almidon agrio in Colombia (Mestres, Zakhia, & Rouau, 1997), is used for production of special kinds of gluten-free breads and biscuits that are very popular in some countries of South America. The traditional process needs long times to produce sour cassava. Therefore many research groups tried to find out more efficient process to produce starch that have baking expansion ability (Demiate, 2000; Mestres et al., 1997; Vatanasuchart et al., 2003 & 2005). Baking expansion related with starch depolymerization by oxidation. Previous studies have proposed a consecutive reaction path in which hydroxyl groups in starch molecules are first oxidized to carbonyl groups and then to carboxyl groups (Kuakpetoon & Wang, 2006) Lactic acid in sour cassava starch can oxidized starch and produced carboxyl groups. UV irradiation could increase carboxyl content of sour cassava starch (Dias et al., 2011). Bertolini et al. (2001) concluded that UV irradiation induced cassava starch depolymerization more effective at 254 nm (UV-C). A recent research also suggests that UV-C irradiation was more effective to enhance baking expansion of cassava starch (Vatanasuchart et al, 2003 & 2005).

Starch also oxidized using oxidizing agent (Kanth et al., 2006; Zhang et al., 2007; Wang & Wang, 2003; Paravuori et al., 1005) Trail Wang, 2003; Paravuori et al., 1995). Hydrogen peroxide, an alternative oxidizing agent, has been used in a commercial practice to been used in a commercial practice to much lesser extent. Hydrogen peroxide creates to harmful byproduct. Hydrogen peroxide descriptions of the commercial practice to much lesser extent. Hydrogen peroxide descriptions of the commercial practice to much lesser extent. harmful byproduct. Hydrogen peroxide decomposes inevitably to oxygen and water (Isbell & Frush, 1987). Hydrogen peroxide was formally to oxygen and water (Isbell & Gridalion and Water (I Frush, 1987). Hydrogen peroxide was faster than hypochlorite for starch oxidation (Sangseethong et al., 2010). In previous study, Harmon et al. (1971 & 1972) had reported the oxidation of starch with hydrogen peroxide that catalyzed by UV irradiation, but they were no relation with baking expansion.

The present study was aim to find out the technique to produce cassava starch that having high baking expansion by oxidation use hydrogen peroxide and UV-C irradiation, in various length of time. The irradiated oxidized cassava starch will be characterized its pH, carboxyl content, baking expansion, swelling and solubility. Irradiated oxidized cassava starch with the highest baking expansion was applied in low gluten bread. Bread would be evaluated its volume and texture.

2. Materials and Methods

2.1 Materials

Commercial cassava starch was obtained from the near store, the chemicals and reagent were both locally.

2.2 Methods

Oxidation process

A slurry containing 42%cassava starchwas added with solution 30% H₂O₂ 0.4% (v/v) and irradiated using UV-C lamp (39 watt, 220-240 V, dimension 83 mm x 900 mm (Alinco product)) in continuous system for 0.25 - 15 min. For the safety reason, this lamp use chamber stainless steel.

Level of pH

Acidity expressed as pH values were measured after suspension of 10% (w/v) starch in distilled water for 30 min under agitation at room temperature. After a further 30 min, starch was decanted and the pH of the soluble fraction was measured (Demiate, 2000).

Carboxyl Content

The carboxyl content of the samples was determined as described by Smith (1967) and employed by Parovuori et al. (1995). Amount of 500 mg of sample (DW) were suspended on 300 ml de-ionized water and boiled for 10 min with constant agitation. These solutions were titrated with NaOH 0.025 M employing phenolphthalein as indicator. Carboxyl content was calculated as shown:

COOH% = ml NaOH x 0.025 x 0.045 * x 100 / 0.5 g(* COOH molecular weight/1000)

Baking expansion

The measurement of baking expansion uses the method of Demiate et al (2000) with a little modification. The baking property was measured by weighing 24 g of sample and partially cooking by addition of 30 ml of boiling de-ionized water over this starch mass. This partially cooked starch was homogenized to produce dough thatwas molded to three small balls and baked on an electric oven at 200°C for 25 min. After baking, the dough was weighed, and made impermeable by using paraffin and their volumes determined on graduated cylinders as the volume of water displaced. The expansion was obtained by dividing volume by weight and was expressed as specific volume (ml/g).

Swelling and Solubility

Swelling power and solubility were determined following method of Lorlowhakarn (2005) with slight modification. Starch sample (0.5 g) in 15 ml of distilled water was heated to the desired temperature (65, 75,85, and 95°C) for 30 min in shaking water bath, then centrifuged

at 4000 rpm for 15 min. The supernatant was evaporated at 100°C until its wallen the swollen was evaporated at 100°C until its wallen wallen was evaporated at 100°C until its wallen wallen was evaporat at 4000 rpm for 15 min. The supernatant was evaporated at 100°C until its weight

Bread preparation
The preparation started by weighing all ingredients. The following step was the mixture and preparation started by weighing all ingredients. The following step was the mixture and sequence, adding first the dry ingredients, followed in a sequence. The preparation started by weighing all ingredients. The preparation started by weighing all ingredients. The preparation started by weighing all ingredients. The dry ingredients, followed by the kneading, according to a specified sequence, adding first the dry ingredients, followed by the kneading, according to a specified sequence, adding many speed experimentally determined egg, yeast, water, and margarine, according to times and speed experimentally determined egg, yeast, water, and margarine, according to thines and the service of the serv After homogenization, 520 g of bread dough were transfer was carried out in electric oven incubated at room temperature in 2 hours. The baking process was carried out in electric oven incubated at room temperature and were cooled down at room temperature and incubated at room temperature in 2 hours. The baked bread were cooled down at room temperature and then wrapped up in plastic bags.

Bread volume and texture

Bread volume and texture

The volume of the thin loaf was evaluated by measurement of length x width x height of the formula: specific volume (a shift of the formula). The volume of the thin loar was evaluated by the formula: specific volume (cm³/g) = bread (cm³). Specific volume was determined using the formula: specific volume (cm³/g) = volume (cm³) / dough (g). Bread texture was measured as the resistance to an applied shear force using Universal Testing Machine (UTM). Force required to attain a given deformation was measured as F max (in Newton).

3. Results and Discussion

3.1 Level of pH and carboxyl content

Carboxyl content is parameter of starch oxidation level. The higher carboxyl content means the higher level of starch oxidation. The presence of carboxyl groups would influence the pH value of starch. Figure 1 exhibited that oxidation and UV-C irradiation decrease pH value of cassava starch after 4 min irradiation (6.19-5.22). In contrary, UV-C irradiation for 0.25-3 min caused increase pH value (6.60 - 6.81). Oxidized cassava starch after irradiated by UV-C for 15 min has the lowest pH value (5.22) and the highest carboxyl content (0.428%). The relationship between the formation of carboxyl groups during starch oxidation and pH value is not yet completely understood. Previous studies have proposed a consecutive reaction path in which hydroxyl groups in starch molecules are first oxidized to carbonyl groups and then to carboxyl groups (Kuakpetoon & Wang, 2006). It can be assumed that in earlier stage of starch oxidation, many of carbonyl groups wasproduced, so the pH value getting higher than

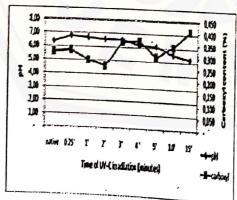


Figure 1 pH and carboxyl content of native and irradiated oxidized cassava starches

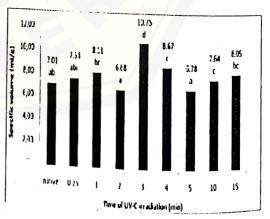


Figure 2 Baking expansion of native and irradiated oxidized cassava starches

3,2 Baking Expansion

Starch oxidation and depolymerization could enhance baking expansion of cassava starch. Baking expansion expressed as specific volume of baked product of cassava starch. Figure 2 shows the specific volumes of cassava starch that oxidized with H₂O₂ 0.4% (v/v) and irradiated by UV-C in different time. The highest specific volumes (10.75 mL/g) were observed within an optimum irradiation range (at 3 min). Our results are consistent with those from Vatanasuchart et al. (2003 & 2005) who obtained the highest specific volumes of irradiated cassava starch on the optimum time of UV irradiation. Interestingly, the highest specific volume was not obtained on irradiated oxidized cassava starch with the highest value of carboxyl groups.

This result relevance with Dias et al work (2011) that conclude that baking expansion ability of oxidized cassava starch was influenced by optimum ratio carbonyl and carboxyl. This behaviour is likely due to not only required carboxyl groups to obtain maximum baking expansion. Demiate et al. (2000) observed that there is a great contribution of the carboxylate group associated with some others difference in starch at around 1060cm-1 on spectral region. After 5 min UV-C irradiated, oxidized cassava starch had lower specific volume (6.78 mL/g) than nativee (7.01 mL/g).

It is possible that this negative effect on the expansion capacity is due to excessive oxidation and increased starch degradation (from breaking glycosidic linkages and increased depolymerisation), which affects the ability of the starch to form internal alveolar structures responsible for expansion. When degradation was too extensive the bubble walls lost their integrity earlier, they ruptured at lower strain causing no expansion of the starches.

3.3 Swelling and Solubility

Table 1 Swelling and solubility of native and irradiated oxidized cassava starches

Samples		Swelling (g/g)		
३र° ६००० च्या च्या च्या च्या च्या च्या च्या च्या	65°C	75°C	85°C	95°C
Native	2.630	5.322	6.652	4.827
0.4% H ₂ O ₂ UV-C 0.25 min	2.450	4.632	4,974	4.308
0.4% H ₂ O ₂ UV-C 1 min	2.311	4.182	4.989	3.513
0.4% H ₂ O ₂ UV-C 2 min	2.354	4.562	4.418	4.133
0.4% H ₂ O ₂ UV-C 3 min	2.098	4.257	4.899	4.461
0.4% H ₂ O ₂ UV-C 4 min	1.853	4.450	5.411	6.472
0.4% H ₂ O ₂ UV-C 5 min	2.900	4,722	3.473	3.510
0.4% H ₂ O ₂ UV-C 10 min	2.892	5.326	4.157	3.655
0.4% H ₂ O ₂ UV-C 15 min	2.344	5.834	4.689	3.708

Table 1 Swelling and solubility of native and irradiated oxidized cassava starches (Cont.)

diste 1 Swelling and se	DIAX XXIII	Sol	unity (70)		
Samples	65°C	75°C	85°C	95°C	
	0.000	0.265	0.349	0.604	
Native O AN C 0 25 min	0.035	0.273	0.304	0.479	<u> </u>
0.4% H ₂ O ₂ UV-C 0.25 min	0.000	0.086	().282	0.417	_
0.4% H ₂ O ₂ UV-C 1 min	0.107	0.247	0.351	0.429	_
0.4% H ₂ O ₂ UV-C 2 min 0.4% H ₂ O ₂ UV-C 3 min	0.035	0.249	0.290	0.348	
0.4% H ₂ O ₂ UV-C 4 min	0.105	0.084	0.305	0.632	_
0.4% H ₂ O ₂ UV-C 5 min	0.113	0.162	0.357	0.301	
0.4% H ₂ O ₂ UV-C 10 min	0.220	0.164	0.407	0.265	
0.4% H ₂ O ₂ UV-C 15 min	0.144	0.454	0.555	0.423	

Oxidized cassava starch that irradiated UV-C in 3 min has the highest baking expansion (10.75 mL/g) and swelling value was lower (4.899 g/g) than native (6.652 g/g) (Table 1). It has been suggested that expansion related to starch swelling (Mestres et al., 2000) attributable of amylopectin. At the beginning of baking, the starch gelatinization induces its partial solubilization and swelling and leads the formation of viscoelastic material. This latter can trap the gas being produced during baking, mainly before the formation of a rigid crust due to the progressive dehydration and which impede the dough expansion. When degradation was too extensive, disturbance in amorphous phase of starch granules was more, so starch swelling getting lower cause of too many water in starch granule. However, the bubble walls lost their integrity earlier causing no expansion of the starches. The high solubilization of irradiated, oxidized cassava starchwas attributed to higher amylopectin solubilization (Mestres, Zakhia, Dufour, 1997). Oxidation by H₂O₂ and UV-C irradiation leads the degradation in amylopectin molecules (Bertolini et al, 2001) which could impact the decreasing swelling and solubility.

3.4 Bread properties and texture

Irradiated oxidized cassava starch (oxidized cassava starch that irradiated by UV-C at 3 min) which highest baking expansion was applied to produce low gluten bread. As a control is bread prepared with wheat flour which exhibited the best volume (3.437cm3/g) and appearance. It was interesting that bread was made from 50% irradiated oxidized cassava starch has a slight volume difference (3.268 cm3/g), only 5% lower than wheat bread volume. Texture can be considered a manifestation of the rheological properties of a food and constitutes an important quality attribute, since it has great influence on the judgment of bread acceptability (Clerici et al., 2009). The resistance of bread crumb to deformation is the is strongly correlated with consumers' perception of bread freshness (Ahlborn et al., 2005). In contrast, baked product with a higher baking expansion usually needed a lower force to break it (Saputro, 2010). In this work, compared with wheat bread, irradiated oxidized cassava starch bread had more crispy texture. This result was relevance

with earlier statement, that cassava bread and wheat bread had difference mechanism for making expansion during baking (Bertolini, et al., 2001). The difference mechanism could influence difference crumb bread properties and texture.

4. Conclusion

The longer time of UV-C irradiation increase baking expansion ability of oxidized cassava starch until initial time. Treatment of cassava starch by adding H₂0₂ 0.4% (v/v) and UV-C irradiation for 3 min resulted in oxidized starch having characteristics: pH 6.59, carboxyl content 0.360%, highest baking expansion ability (specific volume 10.75 ml/g), highest swelling power at 85°C (4.899 g/g), highest solubility at 95°C (0.348%). Thismethod was more efficient than that of previous study because require short time to produce cassava starch that has high baking expansion. Bread prepared with irradiated oxidized cassava starch (50%) not only produce a high volume bread and need a low force to break it, but also had a good appearance. It can be accepted that irradiated oxidized cassava starch can be substituted wheat to produce low gluten bread.

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